## COLOURED METHS LOCAL SALES SPECIFICATION

DESCRIPTION				
A blend of C2 and C3 alcohol	s			
SPECIFICATION				
PROPERTIES	UNITS	LIMITS	TEST METHOD	
Ethanol content	Mass %	91 min	GC	
Subrem				

OTHER PHYSICAL PROPERTIES				
PROPERTIES	UNITS	TYPICAL VALUES	TEST METHOD	
Distillation Range Initial boiling point 50% Final boiling point	~ °C	75 78 118	ASTM D1078	
Boiling Point – Recovery up to 78℃	Vol %	79		
Boiling Point – Recovery up to 80 ℃	Vol %	83		
Density at 20 ℃	Kg/l	<mark>0,785 –</mark> 0,792	ASTM D1298	
Acidity as CH3COOH	Mg/g	0,01	ASTM D1613	
Water content	Mass %	0,4	ASTM D1364	
Flash Point (closed cup) at 101,3 kPa	∞ ~	13	IP 170	
Residue on evaporation	Mg/100ml	1	ASTM D1353	
Evaporation rate, relative	n-BuAc = 1	1,18	ASTM D3539	
Appearance	-	Clear and free of sediment	VISUAL	

Hazchem number : 2SE

**UN number : 1170** 

## Uses:

A solvent in the following applications: adhesives, lacquers, lacquer thinners, etch primers, varnishes, stains, paper coatings, abrasive manufacture, liquid inks, industrial detergents, air fresheners and in latex rubber production.

Product Name	ETHYLOL		
Commercial Product Name	ETHYLOL		
Synonyms	Mineralised methylated spirits, coloured methylated spirits, coloured meths.		
Suprem			

HEALTH HAZARD INFORMATION				
Causes	Symptoms	Emergency and first aid procedures		
Inhalation (Breathing)	Vapour is irritating to nose and throat.	If breathing is affected, remove victim to fresh air, call physician, administer oxygen.		
Skin Contact / Absorption	Not harmful.	Wash with soap and water.		
Eye Contact	Vapour may irritate eyes.	Flush eyes thoroughly with water.		
Ingestion (Swallow)	Headache and drowsiness may occur. Can cause intoxication. It is a central nervous system depressant.	If patient is unconscious, keep warm, obtain immediate medical attention.		
Acute over-exposure	Large doses can cause alcohol poisoning. 400ml, taken within one hour can be deadly.			
Chronic over-exposure	Repeated ingestions can lead to alcoholism.			

HAZARDOUS COMPONENTS		REACTIVITY DATA		
Components	%	Hazard Data	Conditions contributing to instability	Can react vigorously with acetyl chloride, BrF3, Ca
Ethanol	88	TLV = 1000 ppm		(OCI)2, CIO3, CrO3, HNO3,
Iso-Propanol	7	TLV = 200 ppm		H2O2
Butanol	4			
			Incompatibility	None
			Hazardous decomposition	None
			products	
			Conditions contributing to hazardous polymerization	Not pertinent

SPECIAL PROTECTION DATA		SPILL OR LEAK PROCEDURES	
Ventilation requirements	Area must be well ventilated.	Steps to be taken if material is released or spilled:	
Respiratory (in detail)	Not pertinent.	Neutralising chemicals	Not pertinent
Eyes	Safety goggles.	Waste disposal method	Mop up with plenty of water and run to waste, diluting
Gloves	Rubber gloves.	rem	greatly with running water. Ventilate area well to evaporate remaining liquid and dispel vapour.
FIRE AND EXPLOSION DATA		PHYSICAL DATA	
Flash Point (Test Method)	13°C (Closed cup)	Boiling Point (at 760mm Hg)	78℃
Auto-ignition temperature	423℃	Melting Point (pour pt)	-114℃
Flammable limits in air % by Vol	Lower: 3,3% Upper: 19%	Density (at 20 ℃)	0,789 kg/litre
Extinguishing media	Alcohol foam, CO2, dry	Vapour pressure (at 19℃)	40mm Hg
	chemical.	Vapour density (Air = 1)	1,59
Special fire fighting Procedures	None	Solubility in H2O	Highly soluble in water
Unusual fire and explosion hazards	Dangerous when exposed to heat or flame	Appearance and odour	Clear, violet, fragrant liquid burning and bitter after taste.

## **Manufacturers Note:**

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